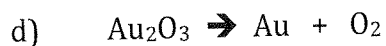
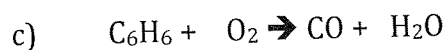
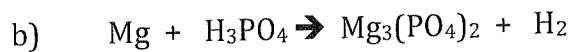
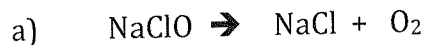


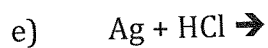
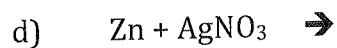
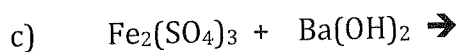
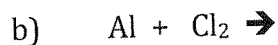
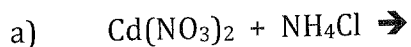
Chapter 11 Test Review

Name: _____

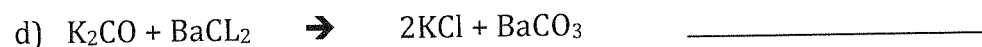
1. Balance the following equations.



2. Complete and balance the following equations. Remember to check the activity series to see if the reaction takes place.



3. **Indicate the type of reaction for each of the following.**



4. Write a balanced equation for the complete combustion of C_4H_8

5. What would be the difference in the above equation if the combustion is not complete?

6. **Will a precipitate form when the following aqueous solutions of ionic compounds are mixed? Indicate what the precipitate would be.**

